

Rubaroc® Softwash

### Rubaroc® Softwash

### Section 01 - Chemical and Product and Company Information

Product Identifier Softwash (Sodium Hypochlorite (5-20%)

Product Use Disinfectant, bleaching agent, source of available chlorine, deodorizer.

Supplier Name Rubaroc

2416 Wyecroft Rd Unit #3 Oakville, ON L6L 6M6

Prepared by Rubaroc Regulatory Department

Phone: 1 (888) 763-7276 **Preparation Date** 08/15/2018

### 24-Hour Emergency Phone

613-966-6666

Section 02 - Com	position /	Inf	ormation	on	Inaredients :

Hazardous Ingredient	Sodium Hypochlorite	4.90-16.80%
CAS Number	Sodium Hypochlorite	7681-52-9

Synonym(s) Industrial bleach, hypo bleach, Javelle water, household bleach

#### Section 03 - Hazard Identification

Inhalation Irritant of the nose and throat, causing coughing, difficulty

breathing, and pulmonary edema.

**Skin Contact / Absorption** Causes severe skin irritation with blistering and ulceration.

**Eye Contact** Causes severe irritation of the mucous membranes of the eyes.

May cause severe eye damage.

Ingestion Burning of the mouth and throat, abdominal cramps, nausea,

vomiting, diarrhea, shock. May lead to convulsions, coma, and

even death.

**Exposure Limits** ACGIH/TLV-TWA: 0.5ppm (chlorine)

### Section 04 - First Aid Measures

Inhalation Remove victim to fresh air. Give artificial respiration only if

breathing has stopped. If breathing is difficult, give oxygen. Seek

immediate medical attention.

Skin Contact / Absorption Remove contaminated clothing. Wash affected area with soap and

water. Seek medical attention if irritation occurs or persists.

Eye Contact Flush immediately with water for at least 20 minutes. Forcibly hold

eyelids apart to ensure complete irrigation of eye tissue. Seek

immediate medical attention.



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**Ingestion** Do not induce vomiting. If vomiting occurs, lean victim forward to

prevent breathing in vomitus. Give large amounts of water. Do not give anything by mouth to an unconscious or convulsing person.

Seek immediate medical attention.

Additional Information Not available

Section 05 - Fire Fighting

Conditions of Flammability Non-flammable

Means of Extinction Product does not burn. Use appropriate extinguishing

media for material that is supplying the fuel to the fire.

Flash Point Not applicable
Auto-ignition Temperature Not applicable

Upper Flammable Limit

Lower Flammable Limit

Not applicable

Not applicable

Hazardous Combustible Products Decomposition may produce chlorine gas and/or hydrogen

chloride gas.

**Special Fire Fighting Procedures** Wear NIOSH-approved self-contained breathing apparatus

and protective clothing.

**Explosion Hazards** Pressure buildup in containers could result in an explosion

when heated or in contact with acidic fumes. Vigorous reaction with oxidizable organic materials may result in a

fire.

#### Section 06 - Accidental Release Measures

**Leak / Spill** Wear appropriate personal protective equipment. Ventilate area.

Stop or reduce leak if safe to do so. Restrict access to spill area until clean-up is complete. Prevent material from entering sewers,

waterways or confined spaces. Soak up smaller spills with

absorbent material that does not react with spilled material. Flush

with water to remove any residue.

**Deactivating Materials** Spills can be carefully neutralized first with sodium sulphite,

sodium metabisulphite or other dichlorination agent for no chlorine residual, then a pH adjustment may be required with hydrochloric acid until the pH is 7. Note neutralization reactions may produce heat so necessary precautions must be taken. Local



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Section 07 - Handling and Storage

Use sensible industrial hygiene and housekeeping practices. Wash thoroughly after handling. Avoid all situations that could lead to

harmful exposure.

Storage Requirements Store in a cool, dry, well-ventilated place. Keep container tightly

closed, and away from incompatible materials. Venting of

containers is advisable.

Section 08 - Personal Protection and Exposure Controls

Protective Equipment

**Eyes**. Chemical goggles, full-face shield, or a full-face respirator is to be

worn at all times when product is handled. Contact lenses should

not be worn; they may contribute to severe eye injury.

Respiratory A NIOSH-approved respirator suitable for chlorine is

recommended. Where a higher level of protection is required, use

a self-contained breathing apparatus.

Gloves Impervious gloves of chemically resistant material (rubber or PVC)

should be worn at all times. Wash contaminated clothing and dry

thoroughly before reuse.

Clothing Body suits, aprons, and/or coveralls of chemical resistant material

should be worn at all times. Wash contaminated clothing and dry

thoroughly before reuse.

Footwear Impervious boots of chemically resistant material should be worn

at all times.

**Engineering Controls** 

Ventilation Requirements Mechanical ventilation (dilution or local exhaust), process or

personnel enclosure and control of process conditions should be provided. Supply sufficient replacement air to make up for air

removed by exhaust systems.

Other Emergency shower and eyewash should be in close proximity.

Section 09 - Physical and Chemical Properties

Physical State Liquid

Odor and Appearance Strong chlorine odour. Clear, greenish-yellow solution.

Odor Threshold Not available

Specific Gravity (Water=1) 1.17 at 20°C (12% trade)



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**Vapor Pressure (mm Hg, 20C)** 12.1mm Hg at 20°C (12.5 wt %)

Vapor Density (Air=1)Not availableEvaporation RateNot available

**Boiling Point** Slowly decomposes above 40°C.

Freeze/Melting Point ~ -15°C (12% trade)

pH < 12

Water/Oil Distribution Coefficient

Bulk Density

Not available

Volatiles by Volume

Solubility in Water

Molecular Formula

Not available

74.44

Section 10 - Stability and Reactivity

Stability Unstable at temperatures above 40°C, in sunlight,

and in contact with acid.

Incompatibility Incompatible with strong acids, ammonia,

oxidizable materials, nickel, copper, tin,

manganese, and iron.

Hazardous Products of Decomposition Chlorine (by reaction with acids), oxygen (by

reaction with nickel, copper, tin, manganese, iron), sodium chloride, sodium chlorate, with increased

temperature.

Polymerization Will not occur

Section 11 - Toxicological Information

IrritancyStrong irritantSensitizationNot available

Chronic/Acute Effects If over-exposed to the solution, there will be constant

irritation of the eyes, nose, and throat.

Synergistic Materials Not available

Animal Toxicity Data LD<sub>50</sub>(oral,rat): 8910mg/kg (undiluted sodium hypochlorite)

Carcinogenicity Not considered to be carcinogenic (IARC and ACGIH).

Reproductive ToxicityNot availableTeratogenicityNot availableMutagenicityNot available



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Section 12 - Ecological Inform	nation
Fish Toxicity	Not available
Biodegradability	Not available
Environmental Effects	Not available
Section 13 - Disposal Consid	eration
Waste Disposal	Dispose in accordance with all federal, provincial, and/or
	local regulations including the Canadian Environmental
	Protection Act.
Section 14 - Transportation In	nformation
TDG Classification	
Class	8 (not regulated at solutions below 7%)
Group	III (not regulated at solutions below 7%)
PIN Number	UN 1791(not regulated at solutions below 7%)
Other	Secure containers (full and/or empty) with suitable hold
	down devises during shipment.
Section 15 - Regulatory Information	mation
WHMIS Classification	E

NOTE: THE PRODUCT LISTED ON THIS MSDS HAS BEEN CLASSIFIED IN ACCORDANCE WITH THE HAZARD CRITERIA OF THE CANADIAN CONTROLLED PRODUCTS REGULATIONS. THIS MSDS CONTAINS ALL INFORMATION REQUIRED BY THOSE REGULATIONS.

#### **NSF** Certification

Product is certified under NSF/ANSI Standard 60 for disinfection and oxidation at a maximum dosage for the following:

sodium hypochlorite 5%: 200mg/L	sodium hypochlorite 13%: 82mg/L
sodium hypochlorite 6%: 175mg/L	sodium hypochlorite 14%: 76mg/L
sodium hypochlorite 7%: 161mg/L	sodium hypochlorite 15%: 70mg/L
sodium hypochlorite 8%: 146mg/L	sodium hypochlorite 16%: 66mg/L
sodium hypochlorite 9%: 131mg/L	sodium hypochlorite 17%: 62mg/L
sodium hypochlorite 10%: 116mg/L	sodium hypochlorite 18%: 58mg/L
sodium hypochlorite 11%: 101mg/L	sodium hypochlorite 19%: 54mg/L
sodium hypochlorite 12%: 87mg/L	sodium hypochlorite 20%: 50mg/L



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**Sanitizer Use:** to obtain 10 liters of a 200 mg/L solution as available chlorine, use 16.7 mL of Hypochlor-12 for each 10 liters of clean, potable water.

#### Section 16 - Other Information

**Note:** The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations.